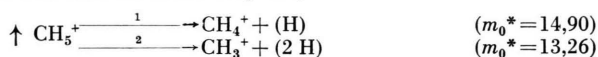


Metastable Dissociation of  $\text{CH}_5^+$  Ions in the Methane Mass Spectrum

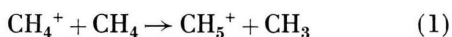
I. MASTAN and V. MERCEA

Atomic Physics Laboratory, Faculty of Physics, Cluj University, Cluj, Rumania

(Z. Naturforsch. **24 a**, 1959—1963 [1969]; received 19 August 1969)The main dissociation channels of the  $\text{CH}_5^+$  ion, i. e.

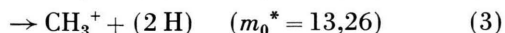
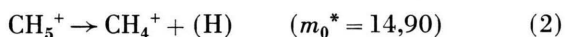
have been investigated by a mass spectrometric method. The pressure dependences of the ratios  $I_{14,90}/I_{17}$  and  $I_{13,26}/I_{17}$ , in the range from  $4,4 \cdot 10^{-4}$  to  $1 \cdot 10^{-4}$  Torr, were plotted. The two ratios depend linearly on the pressure in the mass spectrometer analyzer. Since the two straight lines do not pass through the origine the conclusion is drawn that, for both dissociation channels of the  $\text{CH}_5^+$  ion, besides the collision-induced transitions, spontaneous transitions, i. e. pure metastable transitions, take place.

The spontaneous dissociations constants ( $\lambda_1 = 7,5 \cdot 10^5 \text{ sec}^{-1}$ ,  $\lambda_2 = 5,2 \cdot 10^4 \text{ sec}^{-1}$ ) and collision-induced dissociation cross-sections ( $\sigma_1 = 5,25 \cdot 10^{-16} \text{ cm}^2$ ,  $\sigma_2 = 3,2 \cdot 10^{-16} \text{ cm}^2$ ) have been estimated. These results are discussed and compared with those reported by other authors.

The often investigated  $1^{-11}$  reaction

is not the only possible one between the parent  $\text{CH}_4^+$  ion and the neutral methane molecule, many others, leading to the formation of heavier ions, being known, but it is generally accepted that it has the largest cross-section. Reaction (1) takes place at low energies, especially in the mass spectrometer ion source. At larger energies, corresponding to the whole accelerating voltage of conventional mass spectrometers, dissociation reactions of the parent ion take place <sup>11</sup>.

As to the stability of the  $\text{CH}_5^+$  ion, FIELD et al. <sup>5</sup> investigated the dissociation channels of the  $\text{CH}_5^+$  ion in a mass spectrometer at a pressure of 320 mTorr. The relative cross-sections for the different dissociation processes were identified. From their results it follows that the dissociation processes



have by far the largest cross-sections. It is specified by them that for the transition (2) the ratio  $I_{14,90}/I_{17}$  increases approximately linearly with the pressure in the mass spectrometer analyzer. The conclusion is drawn that this process is collision-induced. The absolute value of the cross-section of process (2) is  $2 \cdot 10^{-16} \text{ cm}^2$ .

The dissociation processes of  $\text{CD}_5^+$ ,  $\text{CH}_5^+$  and  $\text{CD}_4^+$  were studied by KUPRIANOV <sup>12</sup> with two mass spectrometers in tandem. Between the two a collision chamber was inserted <sup>13</sup>, in which there was a controlled low pressure atmosphere of neon or air. From the obtained results the author drew the conclusion that in all cases the dissociation of these ions takes place as a result of their collisions with the gas molecules. He concluded that the  $\text{CD}_5^+$ ,  $\text{CH}_5^+$  and  $\text{CD}_4^+$  ions are stable with regard to spontaneous dissociation (i. e. pure metastable transitions), for all the studied reaction channels.

In the present paper the  $\text{CH}_5^+$  ion transitions (2) and (3) were again studied.

<sup>1</sup> F. H. FIELD, J. L. FRANKLIN, and F. W. LAMPE, J. Am. Chem. Soc. **79**, 2419 [1957].

<sup>2</sup> C. D. WAGNER, P. A. WADSWORTH, and D. P. STEVENSON, J. Chem. Phys. **28**, 517 [1958].

<sup>3</sup> V. L. TALROZE and E. L. FRANKOVICH, Zh. Fiz. Khim. **34**, 2709 [1960].

<sup>4</sup> A. HENGLEIN and G. A. MUCCINI, Z. Naturforsch. **17 a**, 452 [1962]; **18 a**, 753 [1963].

<sup>5</sup> F. H. FIELD, J. L. FRANKLIN, and M. S. B. MUNSON, J. Am. Chem. Soc. **85**, 3575 [1963].

<sup>6</sup> G. A. DERWISH, A. GALLI, A. GIARDINI-GUIDONI, and G. G. VOLPI, J. Chem. Phys. **40**, 5 [1964].

<sup>7</sup> H. VON KOCH, Arkiv Fysik **28**, 529 [1965].

<sup>8</sup> F. H. FIELD and M. S. B. MUNSON, J. Am. Chem. Soc. **87**, 3289 [1965].

<sup>9</sup> F. P. ABRAMSON and J. H. FUTRELL, J. Chem. Phys. **45**, 1925 [1966].

<sup>10</sup> F. P. ABRAMSON and J. H. FUTRELL, J. Chem. Phys. **46**, 3264 [1967].

<sup>11</sup> A. GIARDINI-GUIDONI and L. FRIEDMAN, J. Chem. Phys. **45**, 937 [1966].

<sup>12</sup> C. E. KUPRIANOV, Zh. Techn. Fiz. **36**, 2161 [1966].

<sup>13</sup> C. E. KUPRIANOV, Zh. Techn. Fiz. **33**, 823 [1963].



## 1. Apparatus

A Dempster type mass spectrometer with electrical scanning of the masses was used. The schematic representation of the apparatus is given in Fig. 1.

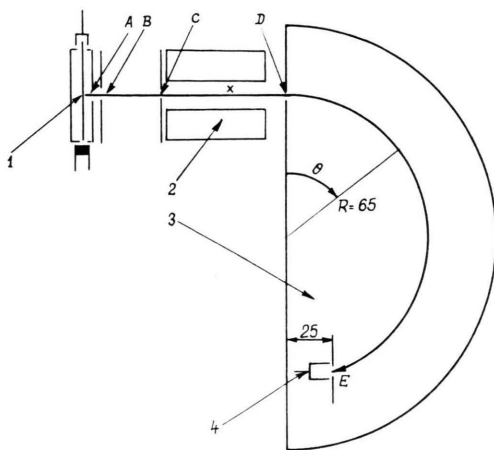
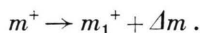


Fig. 1. Schematic representation of the mass spectrometer. 1—beam of ionizing electrons, 2—deflector, 3—analyzer, 4—collector.

A detailed description of the apparatus is given in <sup>14</sup>. Some characteristics which are of interest in the present paper, will be given. The voltages on the slits are: (A)  $\rightarrow$  1–5% of  $V$ ; (B)  $\rightarrow$  25% of  $V$  and (C, D)  $\rightarrow$  100% of  $V$ ,  $V$  being the whole accelerating voltage of the ions (from 360 to 3000 volts). On the deflector plates (corresponding to the trajectory CD) a potential  $\sim 80\%$  of  $V$  is applied. Thus, in the deflector the ions will practically keep the energy gained at point C of the trajectory.

The shape of the metastable “peak” depends on the mass spectrometer characteristics. We give some specifications concerning the shape of our metastable “peaks” in <sup>14</sup>.

The following metastable transitions are considered



The apparent masses  $m^*$  at which the  $m_1^+$  ions will be collected depend on the trajectory points at which the dissociation took place.

If dissociation takes place at A (cf. Fig. 1), then  $m^* = m_1$  and if dissociation takes place at C, then  $m_0^* = m_1^2/m$ . The other apparent masses are in between these limiting values.

Since the slits C and D are at the same potential all ions dissociating on the trajectory portion CD will be practically collected at the same apparent mass  $m_0^* = m_1^2/m$ . At this apparent mass they will form a metastable peak similar to the normal peaks of the mass spectrum. This apparent mass,  $m_0^* = m_1^2/m$ , corresponds to the maximum height of the metastable “peak”.

The ions which are dissociated on the trajectory portion DE will be collected at masses ranging from  $m_0^* = m_1^2/m$  (if dissociation takes place in the neighbourhood of D), to  $m^* = m$  (if dissociation takes place at the collector).

The superposition of the contributions corresponding to the trajectory portions AC, CD and DE will give the shape of the metastable “peak”. Thus, metastable “peaks” in our mass spectrometer are asymmetrical and very diffuse.

The measurements were performed in a range of pressures in the mass spectrometer analyzer from  $4.4 \cdot 10^{-4}$  to  $1 \cdot 10^{-4}$  Torr. The mass spectrometer ion source is of the Nier type. The energy of the ionizing electrons is constant and equal to 85 eV.

## 2. Experimental Data

In a previous paper <sup>14</sup> the methane mass spectrum obtained in our mass spectrometer with a relatively high pressure in the analyzer and the ion source was studied. An approximate estimation of the shapes of the main collision-induced metastable transition “peaks” was given. Using these results we have studied the most important channels of the  $\text{CH}_5^+$  ion dissociation as indicated in (2) and (3). For this purpose we have investigated the pressure dependence of the ratios  $I_{14.90}/I_{17}$  and  $I_{13.26}/I_{17}$ . The obtained results are given in Fig. 2.

The two ratios depend linearly on the pressure in the mass spectrometer analyzer. From the fact that the two straight lines do not pass through the origine we conclude that besides the collision-induced transitions, spontaneous transitions, i. e. pure metastable transitions, take place.

It is likely that for the energy of 85 eV of the ionizing electrons the  $\text{CH}_4^+$  ions have a large internal energy. Hence, a part of the  $\text{CH}_5^+$  ions can have sufficiently large internal energies to undergo spontaneous dissociation on their way from the source to the collector. From Fig. 2 it can be seen that in the spontaneous dissociation of the  $\text{CH}_5^+$  ion the forma-

<sup>14</sup> I. MASTAN and V. MERCEA, to be published.



in the form

$$I_{14,90}^* = (I_a)_{AC} \exp\{-\lambda_1 t_C\} - (I_a)_{CD} \exp\{-\lambda_1 t_D\}. \quad (9)$$

For  $(I_a)_{AC}$  and  $(I_a)_{CD}$  one can write

$$(I_a)_{AC} = (I_0/2) (1 + \Psi_1), \quad (10)$$

$$(I_a)_{CD} = (I_0/2) \Psi_1 (1 + \Psi_2) \quad (11)$$

with

$$\Psi_1 = \exp\{-(\sigma_1 N + \sigma_2 N) (k/2) t_C\},$$

$$\Psi_2 = \exp\{-(\sigma_1 N + \sigma_2 N) k(t_D - t_C)\}.$$

Using (10) and (11) Eq. (9) can be written in the form

$$I_{14,90}^* = (I_0/2) [(1 + \Psi_1) \exp\{-\lambda_1 t_C\} - \Psi_1 (1 + \Psi_2) \exp\{-\lambda_1 t_D\}]. \quad (12)$$

Similarly, for  $I_{13,26}^*$  one can write

$$I_{13,26}^* = (I_0/2) [(1 + \Psi_1) \exp\{-\lambda_2 t_C\} - \Psi_1 (1 + \Psi_2) \exp\{-\lambda_2 t_D\}]. \quad (13)$$

The following notations will be introduced

$$(I_{14,90}^*/I_{17}) = A_1, \quad (I_{13,26}^*/I_{17}) = A_2,$$

$$A_1 Q \exp\{-\lambda_1 t_E\} = R, \quad \sqrt{\frac{1}{2}} [(1 + \Psi_1) \exp\{-\lambda_1 t_C\} - \Psi_1 (1 + \Psi_2) \exp\{-\lambda_1 t_D\}] = P$$

and

$$\lambda_2 = \ln(R/P)/t_E. \quad (14)$$

With these notations the exponential Eqs (7), (12), (13), and (14) lead to the transcendental Equation

$$\frac{A_2 Q \exp\{-\lambda_1 t_E\}}{\frac{1}{2} [(1 + \Psi_1) \exp\{-\lambda_2 t_C\} - \Psi_1 (1 + \Psi_2) \exp\{-\lambda_2 t_D\}]} = \exp\{-\lambda_2 t_E\}. \quad (15)$$

If  $Q$  is known (it implies the knowledge of the values  $\sigma_1$  and  $\sigma_2$ ),  $\lambda_1$  and  $\lambda_2$  can be computed with the help of Eqs (15) and (14).

### 3.2. Estimation of the cross-sections $\sigma_1$ and $\sigma_2$

For that pressure range in which the ratios  $I_{14,90}/I_{17}$  and  $I_{13,26}/I_{17}$  depend linearly on the pressure (cf. Fig. 2), one can write

$$(I_{14,90}/I_{17}) = \sigma_1 p \Delta x + K_1,$$

$$(I_{13,26}/I_{17}) = \sigma_2 p \Delta x + K_2. \quad (16)$$

We specify that by  $I_{17}$ ,  $I_{14,90}$  and  $I_{13,26}$  the whole heights of the peaks are denoted. On the basis of the

<sup>15</sup> In order to avoid any confusions we specify that  $N$  is the concentration expressed in molecules/cm<sup>3</sup>, while  $p$  is the pressure expressed in Torr.

Eq. (4) one can write

$$(I_{14,90}^*/I_{17}) = K_1 = A_1, \quad (I_{13,26}^*/I_{17}) = A_2 = K_2. \quad (17)$$

Hence

$$[(I_{14,90})_{c.i}/I_{17}] = \sigma_1 p \Delta x, \quad [(I_{13,26})_{c.i}/I_{17}] = \sigma_2 p \Delta x. \quad (18)$$

If in (18) one puts  $\Delta x = x_E$  then it is necessary to replace, on the left hand side of these relations, the ratio of the heights with the ratio of the peak areas. We recall that metastable "peaks" given by our mass spectrometer are asymmetrical and very diffuse<sup>14</sup>. For example, the "tail" of the metastable "peak"  $m_0^* = 14,90$  is practically extended up to  $m^* = 17$ . We have approximated with triangles the "peaks" 14,90, 13,26 and 17. By  $\gamma_1$  and  $\gamma_2$  we have denoted the ratios of the bases of these triangles for the peaks 14,90 and 17, and 13,26 and 17 respectively. As a result of these specifications the relations (18) can be written

$$\gamma_1 [(I_{14,90})_{c.i}/I_{17}] = \sigma_1 p x_E, \quad (19)$$

$$\gamma_2 [(I_{13,26})_{c.i}/I_{17}] = \sigma_2 p x_E.$$

For the pressure range in which these ratios are given (cf. Fig. 2) one can write

$$\gamma_1 \cdot \Delta [(I_{14,90})_{c.i}/I_{17}] / (\Delta p \cdot x_E) = \sigma_1,$$

$$\gamma_2 \cdot \Delta [(I_{13,26})_{c.i}/I_{17}] / (\Delta p \cdot x_E) = \sigma_2. \quad (20)$$

With the numerical values  $\gamma_1 = 1,7$ ,  $\gamma_2 = 2,65$  and  $x_E = 21,4$  cm one obtains

$$\sigma_1 = 5,25 \cdot 10^{-16} \text{ cm}^2, \quad \sigma_2 = 3,20 \cdot 10^{-16} \text{ cm}^2.$$

On the basis of the mass spectrometer features (geometry, accelerating voltage, etc.) one gets for the  $\text{CH}_5^+$  ion

$$t_C = 2 \cdot 10^{-7} \text{ sec}, \quad t_E = 1,58 \cdot 10^{-6} \text{ sec},$$

$$t_D = 4 \cdot 10^{-7} \text{ sec}, \quad k = 1,4 \cdot 10^7 \text{ cm/sec}.$$

Using these numerical values, for  $p = 4,4 \cdot 10^{-4}$  Torr, one obtains  $Q = 7,6 \cdot 10^{-1}$ .

### 3.3. Estimation of the constants $\lambda_1$ , $\lambda_2$ and of the $(I_0)_{17}$

Equation (15), where  $\lambda_2$  is given by Eq. (14), is solved using the following known values ( $p = 4,4 \cdot 10^{-4}$  Torr)

$$t_C = 2 \cdot 10^{-7} \text{ sec}, \quad Q = 7,6 \cdot 10^{-1},$$

$$t_D = 4 \cdot 10^{-7} \text{ sec}, \quad A_1 = 13,6 \cdot 10^{-2},$$

$$t_E = 1,58 \cdot 10^{-6} \text{ sec}, \quad A_2 = 9 \cdot 10^{-3},$$

One obtains

$$\lambda_1 = 7,5 \cdot 10^5 \text{ sec}^{-1}, \quad \lambda_2 = 5,2 \cdot 10^4 \text{ sec}^{-1}.$$

With the help of  $\lambda_1$  and  $\lambda_2$  the lifetimes and half-lives for both spontaneous dissociations may be computed

$$\begin{aligned}\tau_1 &= (1/\lambda_1) = 1,33 \cdot 10^{-6} \text{ sec}, \\ \tau_2 &= (1/\lambda_2) = 1,93 \cdot 10^{-5} \text{ sec}, \\ (T_{1/2})_1 &= (\ln 2/\lambda_1) = 0,93 \cdot 10^{-6} \text{ sec}, \\ (T_{1/2})_2 &= (\ln 2/\lambda_2) = 1,33 \cdot 10^{-5} \text{ sec}.\end{aligned}$$

From Eq. (7), using the values obtained here,  $(I_0)_{17}$  can be computed. For  $p = 4,4 \cdot 10^{-4}$  Torr, knowing that  $(I_{17})_E = 32$ , one obtains  $(I_0)_{17} = 142$  (in arbitrary units).

#### 4. Discussion of the Results

A value of  $\sigma_1 = 5,25 \cdot 10^{-16} \text{ cm}^2$  was estimated by us for the cross-section of the collision-induced dissociation corresponding to the reaction



FIELD et al.<sup>5</sup> reported, for the same dissociation process, a cross-section of  $2 \cdot 10^{-16} \text{ cm}^2$ . The agreement between the two values is acceptable if it is taken into account that our determination is only approximative.

In his paper KUPRIANOV<sup>12</sup> specifies that the time necessary for the ions to cross the first mass spectrometer analyzer is of the order of  $6 \cdot 10^{-6} \text{ sec}$ . Under these conditions the fact that one does not observe spontaneous dissociation of the  $\text{CH}_5^+$  ions in the tandem collision chamber<sup>13</sup>, becomes explicable.

For the spontaneous dissociation



which is much more probable than that into  $\text{CH}_3^+$ , we have obtained  $(T_{1/2})_1 = 0,93 \cdot 10^{-6} \text{ sec}$ . If one compares this value with the time of  $6 \cdot 10^{-6} \text{ sec}$  necessary for the ions to reach the collision chamber, it becomes clear that the spontaneous dissociations which occur in the chamber, will be masked by the collision-induced ones.

For the spontaneous dissociation



which is more than 10 times less probable than the first one, we have obtained  $(T_{1/2})_2 = 1,33 \cdot 10^{-5} \text{ sec}$ . Having in view the smallness of this transition probability and the fact that  $(T_{1/2})_2$  is comparable with  $6 \cdot 10^{-6} \text{ sec}$ , it will be difficult to observe it in the collision chamber.